

# (*E*)-*N*-(3,3-Diphenylallylidene)naphthalen-1-amine

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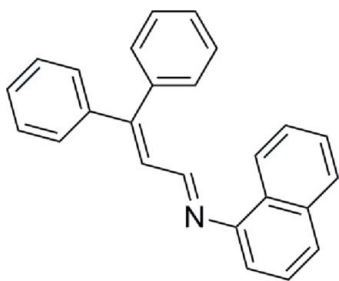
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Key indicators: single-crystal X-ray study;  $T = 296$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å;  $R$  factor = 0.061;  $wR$  factor = 0.197; data-to-parameter ratio = 17.8.

The title compound,  $\text{C}_{25}\text{H}_{19}\text{N}$ , adopts an *E* conformation about the  $\text{C}=\text{N}$  bond. The naphthalene ring system and the phenyl rings form dihedral angles 38.1 (1), 46.9 (8) and 48.5 (1)°, respectively, with the mean plane of the central enimine fragment. The crystal packing exhibits no directional close contacts.

## Related literature

For the crystal structures of related compounds studied recently by our group, see: Cha *et al.* (2012); Kang *et al.* (2012); Yu *et al.* (2013); Nam *et al.* (2013).



## Experimental

### Crystal data

$\text{C}_{25}\text{H}_{19}\text{N}$	$V = 3682.7$ (4) Å <sup>3</sup>
$M_r = 333.41$	$Z = 8$
Orthorhombic, <i>Pbca</i>	Mo $K\alpha$ radiation
$a = 11.2203$ (7) Å	$\mu = 0.07$ mm <sup>-1</sup>
$b = 13.5658$ (7) Å	$T = 296$ K
$c = 24.1946$ (13) Å	$0.20 \times 0.20 \times 0.20$ mm

### Data collection

Rigaku R-Axis RAPID diffractometer	33578 measured reflections
Absorption correction: multi-scan ( <i>ABSCOR</i> ; Rigaku, 1995)	4190 independent reflections
$T_{\min} = 0.986$ , $T_{\max} = 0.986$	2080 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.070$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.061$	235 parameters
$wR(F^2) = 0.197$	H-atom parameters constrained
$S = 0.95$	$\Delta\rho_{\max} = 0.16$ e Å <sup>-3</sup>
4190 reflections	$\Delta\rho_{\min} = -0.25$ e Å <sup>-3</sup>

Data collection: *RAPID-AUTO* (Rigaku, 2006); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *CrystalStructure* (Rigaku, 2010); software used to prepare material for publication: *CrystalStructure*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV5411).

## References

- Cha, J. H., Kang, Y. K., Cho, Y. S., Lee, J. K. & Woo, J. C. (2012). *Acta Cryst. E* **68**, o3030.  
 Kang, Y. K., Cho, Y. S., Lee, J. K., Yu, B.-Y. & Cha, J. H. (2012). *Acta Cryst. E* **68**, o3031.  
 Nam, K. D., Cha, J. H., Cho, Y. S., Lee, J. K. & Pae, A. N. (2013). *Acta Cryst. E* **69**, o548.  
 Rigaku (1995). *ABSCOR*. Rigaku Corporation, Tokyo, Japan.  
 Rigaku (2006). *RAPID-AUTO*. Rigaku Corporation, Tokyo, Japan.  
 Rigaku (2010). *CrystalStructure*. Rigaku Corporation, Tokyo, Japan.  
 Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.  
 Yu, B.-Y., Lee, J. E., Cho, Y. S., Cha, J. H. & Lee, J. K. (2013). *Acta Cryst. E* **69**, o757.

## supplementary materials

*Acta Cryst.* (2013). E69, o1025 [doi:10.1107/S1600536813014888]

**(*E*)-*N*-(3,3-Diphenylallylidene)naphthalen-1-amine**

**Jae Kyun Lee, Kee Dal Nam, Joo Hwan Cha, Yong Seo Cho and Joon Kyun Lee**

**Comment**

As a part of our ongoing study of 2-phenylcinnamaldehyde derivatives containing aniline (Cha *et al.*, 2012; Kang *et al.*, 2012; Yu *et al.*, 2013) and naphthylamine (Nam *et al.*, 2013), we present here the title compound.

The title compound (Fig. 1) adopts an (*E*) conformation about the C=N bond. The naphthalene bicycle C1–C10 and phenyl rings C14–C19 and C20–C25 form the dihedral angles 38.1 (1), 46.9 (8) and 48.5 (1)°, respectively, with the mean plane of the central N1–C11–C12 enmino fragment. The crystal packing exhibits no classical intermolecular contacts.

**Experimental**

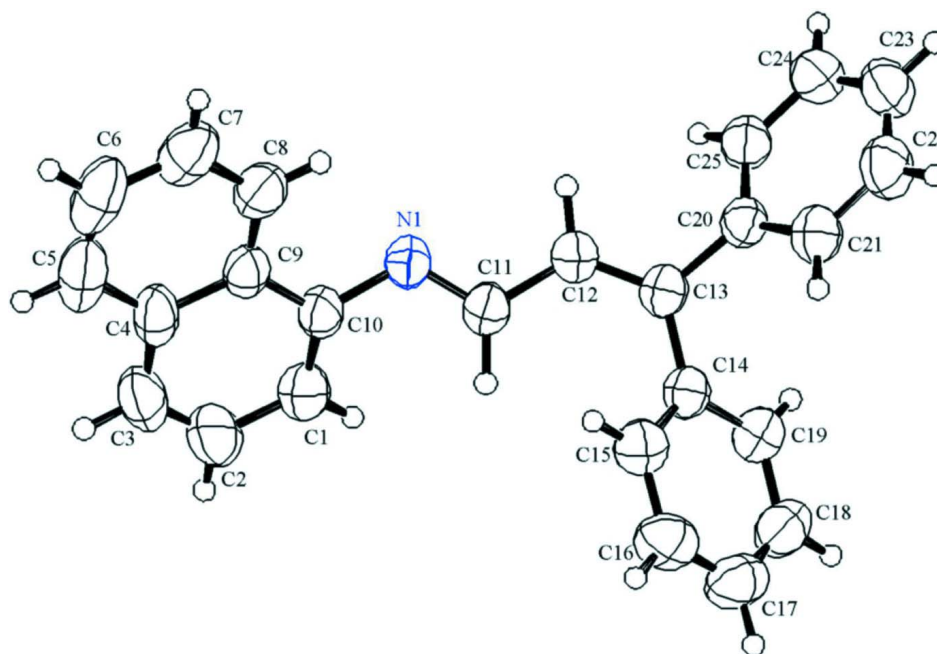
In a solution of 1-naphthylamine (2.0 mmol) in anhydrous ethanol (50 mL) was treated with equimolar quantities of substituted 2-phenylcinnamaldehydes. The mixture was refluxed for 2 days, and the progress of reaction was monitored by TLC. After completion of reaction, the solvent was removed under reduced pressure. The residue was purified by flash column chromatography to afford the title compound as a yellow solid in yield 92%. Single crystals suitable for X-ray diffraction were prepared by slow evaporation of a solution of the title compound in ethanol at room temperature.

**Refinement**

All hydrogen atoms were positioned geometrically (C—H = 0.93 Å), and refined using a riding model, with  $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$ .

**Computing details**

Data collection: RAPID–AUTO (Rigaku, 2006); cell refinement: RAPID–AUTO (Rigaku, 2006); data reduction: RAPID–AUTO (Rigaku, 2006); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *CrystalStructure* (Rigaku, 2010); software used to prepare material for publication: *CrystalStructure* (Rigaku, 2010).



**Figure 1**

The molecular structure of the title compound showing the atomic numbering and 50% probability displacement ellipsoid.

**(*E*)-*N*-(3,3-Diphenylallylidene)naphthalen-1-amine**

*Crystal data*

$C_{25}H_{19}N$

$M_r = 333.41$

Orthorhombic, *Pbca*

Hall symbol: -P 2ac 2ab

$a = 11.2203 (7) \text{ \AA}$

$b = 13.5658 (7) \text{ \AA}$

$c = 24.1946 (13) \text{ \AA}$

$V = 3682.7 (4) \text{ \AA}^3$

$Z = 8$

$F(000) = 1408$

$D_x = 1.203 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71075 \text{ \AA}$

$\mu = 0.07 \text{ mm}^{-1}$

$T = 296 \text{ K}$

Block, yellow

$0.20 \times 0.20 \times 0.20 \text{ mm}$

*Data collection*

Rigaku R-Axis RAPID

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution:  $10.000 \text{ pixels mm}^{-1}$

$\omega$  scans

Absorption correction: multi-scan

(*ABSCOR*; Rigaku, 1995)

$T_{\min} = 0.986$ ,  $T_{\max} = 0.986$

33578 measured reflections

4190 independent reflections

2080 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.070$

$\theta_{\max} = 27.4^\circ$ ,  $\theta_{\min} = 3.0^\circ$

$h = -14 \rightarrow 14$

$k = -17 \rightarrow 17$

$l = -31 \rightarrow 30$

*Refinement*

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.061$

$wR(F^2) = 0.197$

$S = 0.95$

4190 reflections

235 parameters

0 restraints

Primary atom site location: structure-invariant  
direct methods  
Secondary atom site location: difference Fourier  
map  
Hydrogen site location: inferred from  
neighbouring sites

H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.1186P)^2]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.16 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.25 \text{ e } \text{\AA}^{-3}$

### Special details

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted R-factor wR and goodness of fit S are based on  $F^2$ , conventional R-factors R are based on F, with F set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

### Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.03605 (16)	0.37399 (12)	0.86987 (6)	0.0588 (5)
C1	-0.1297 (2)	0.27203 (18)	0.83686 (9)	0.0688 (6)
H1	-0.1143	0.2251	0.8640	0.083*
C2	-0.2224 (3)	0.2562 (2)	0.79886 (10)	0.0830 (8)
H2	-0.2682	0.1992	0.8013	0.100*
C3	-0.2458 (2)	0.3231 (2)	0.75873 (9)	0.0827 (8)
H3	-0.3079	0.3118	0.7341	0.099*
C4	-0.1772 (2)	0.4100 (2)	0.75384 (8)	0.0682 (7)
C5	-0.1976 (3)	0.4814 (3)	0.71211 (10)	0.0888 (9)
H5	-0.2569	0.4705	0.6860	0.107*
C6	-0.1321 (3)	0.5650 (3)	0.70969 (10)	0.0947 (10)
H6	-0.1471	0.6108	0.6819	0.114*
C7	-0.0421 (3)	0.5836 (2)	0.74842 (11)	0.0850 (8)
H7	0.0018	0.6416	0.7465	0.102*
C8	-0.0190 (2)	0.51646 (18)	0.78898 (9)	0.0690 (7)
H8	0.0406	0.5295	0.8147	0.083*
C9	-0.0839 (2)	0.42754 (17)	0.79262 (8)	0.0582 (6)
C10	-0.06209 (19)	0.35522 (16)	0.83444 (8)	0.0564 (5)
C11	0.0290 (2)	0.35090 (15)	0.92139 (8)	0.0558 (5)
H11	-0.0394	0.3213	0.9354	0.067*
C12	0.1286 (2)	0.37159 (14)	0.95689 (8)	0.0562 (5)
H12	0.1979	0.3923	0.9393	0.067*
C13	0.13445 (18)	0.36485 (14)	1.01268 (8)	0.0522 (5)
C14	0.03384 (19)	0.33393 (15)	1.04830 (8)	0.0534 (5)
C15	-0.0793 (2)	0.37540 (16)	1.04349 (9)	0.0630 (6)
H15	-0.0937	0.4227	1.0165	0.076*
C16	-0.1706 (2)	0.3467 (2)	1.07863 (10)	0.0761 (7)
H16	-0.2457	0.3750	1.0751	0.091*
C17	-0.1506 (3)	0.2765 (2)	1.11871 (10)	0.0793 (8)
H17	-0.2122	0.2571	1.1420	0.095*
C18	-0.0399 (3)	0.23538 (18)	1.12413 (9)	0.0747 (7)

H18	−0.0264	0.1881	1.1513	0.090*
C19	0.0517 (2)	0.26348 (16)	1.08965 (8)	0.0642 (6)
H19	0.1266	0.2351	1.0939	0.077*
C20	0.24624 (19)	0.39592 (14)	1.04092 (8)	0.0522 (5)
C21	0.2411 (2)	0.45116 (16)	1.08939 (8)	0.0655 (6)
H21	0.1676	0.4635	1.1058	0.079*
C22	0.3434 (3)	0.48772 (19)	1.11332 (10)	0.0760 (7)
H22	0.3385	0.5257	1.1452	0.091*
C23	0.4527 (3)	0.46821 (19)	1.09016 (10)	0.0781 (7)
H23	0.5216	0.4933	1.1062	0.094*
C24	0.4600 (2)	0.41132 (18)	1.04309 (10)	0.0702 (6)
H24	0.5340	0.3969	1.0279	0.084*
C25	0.3577 (2)	0.37575 (16)	1.01848 (8)	0.0604 (6)
H25	0.3634	0.3379	0.9865	0.072*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
N1	0.0562 (11)	0.0734 (12)	0.0467 (9)	0.0012 (9)	−0.0046 (8)	0.0029 (8)
C1	0.0738 (16)	0.0759 (15)	0.0568 (12)	−0.0053 (13)	−0.0020 (11)	−0.0047 (11)
C2	0.0819 (19)	0.0996 (19)	0.0675 (15)	−0.0148 (15)	−0.0064 (14)	−0.0201 (14)
C3	0.0614 (16)	0.127 (2)	0.0593 (13)	0.0046 (16)	−0.0104 (12)	−0.0266 (15)
C4	0.0574 (15)	0.1024 (18)	0.0447 (11)	0.0215 (13)	−0.0033 (10)	−0.0105 (12)
C5	0.080 (2)	0.133 (3)	0.0530 (13)	0.042 (2)	−0.0055 (13)	−0.0045 (15)
C6	0.108 (3)	0.119 (2)	0.0568 (15)	0.052 (2)	0.0038 (15)	0.0169 (16)
C7	0.087 (2)	0.0966 (18)	0.0716 (15)	0.0220 (15)	0.0105 (14)	0.0225 (14)
C8	0.0638 (15)	0.0851 (17)	0.0580 (13)	0.0107 (13)	0.0052 (11)	0.0094 (12)
C9	0.0501 (13)	0.0809 (15)	0.0436 (10)	0.0106 (11)	0.0032 (9)	−0.0022 (10)
C10	0.0531 (13)	0.0728 (14)	0.0432 (10)	0.0037 (11)	−0.0006 (9)	−0.0030 (9)
C11	0.0597 (13)	0.0604 (12)	0.0473 (10)	−0.0005 (10)	−0.0016 (9)	0.0012 (9)
C12	0.0552 (13)	0.0609 (13)	0.0524 (11)	−0.0005 (10)	−0.0043 (10)	0.0020 (9)
C13	0.0575 (13)	0.0496 (11)	0.0494 (10)	0.0035 (9)	−0.0017 (9)	−0.0002 (9)
C14	0.0587 (13)	0.0538 (11)	0.0477 (10)	0.0004 (10)	−0.0015 (9)	−0.0044 (9)
C15	0.0639 (15)	0.0627 (13)	0.0623 (12)	0.0056 (11)	0.0012 (11)	−0.0043 (10)
C16	0.0628 (16)	0.0831 (16)	0.0824 (16)	0.0050 (13)	0.0132 (13)	−0.0138 (14)
C17	0.083 (2)	0.0854 (18)	0.0691 (14)	−0.0109 (15)	0.0213 (14)	−0.0095 (13)
C18	0.0897 (19)	0.0763 (16)	0.0580 (13)	−0.0081 (14)	0.0090 (13)	0.0040 (11)
C19	0.0681 (15)	0.0682 (13)	0.0563 (12)	−0.0009 (11)	−0.0006 (11)	0.0054 (10)
C20	0.0568 (13)	0.0519 (11)	0.0480 (10)	−0.0005 (10)	−0.0030 (9)	0.0045 (8)
C21	0.0706 (15)	0.0692 (14)	0.0568 (12)	−0.0008 (12)	−0.0029 (11)	−0.0056 (10)
C22	0.089 (2)	0.0747 (16)	0.0643 (14)	−0.0092 (14)	−0.0104 (14)	−0.0122 (12)
C23	0.0777 (19)	0.0796 (16)	0.0770 (16)	−0.0184 (14)	−0.0201 (14)	0.0043 (13)
C24	0.0618 (15)	0.0793 (16)	0.0696 (14)	−0.0057 (12)	−0.0055 (12)	0.0098 (12)
C25	0.0607 (14)	0.0658 (13)	0.0547 (12)	0.0016 (11)	−0.0020 (10)	0.0003 (10)

*Geometric parameters (Å, °)*

N1—C11	1.288 (2)	C13—C14	1.481 (3)
N1—C10	1.419 (3)	C13—C20	1.489 (3)
C1—C10	1.361 (3)	C14—C15	1.394 (3)

C1—C2	1.405 (3)	C14—C19	1.398 (3)
C1—H1	0.9300	C15—C16	1.387 (3)
C2—C3	1.355 (4)	C15—H15	0.9300
C2—H2	0.9300	C16—C17	1.377 (4)
C3—C4	1.412 (4)	C16—H16	0.9300
C3—H3	0.9300	C17—C18	1.368 (4)
C4—C5	1.417 (4)	C17—H17	0.9300
C4—C9	1.425 (3)	C18—C19	1.378 (3)
C5—C6	1.353 (4)	C18—H18	0.9300
C5—H5	0.9300	C19—H19	0.9300
C6—C7	1.400 (4)	C20—C25	1.391 (3)
C6—H6	0.9300	C20—C21	1.393 (3)
C7—C8	1.364 (3)	C21—C22	1.379 (3)
C7—H7	0.9300	C21—H21	0.9300
C8—C9	1.412 (3)	C22—C23	1.373 (3)
C8—H8	0.9300	C22—H22	0.9300
C9—C10	1.430 (3)	C23—C24	1.378 (3)
C11—C12	1.437 (3)	C23—H23	0.9300
C11—H11	0.9300	C24—C25	1.380 (3)
C12—C13	1.355 (3)	C24—H24	0.9300
C12—H12	0.9300	C25—H25	0.9300
C11—N1—C10	119.57 (18)	C12—C13—C20	118.61 (19)
C10—C1—C2	120.7 (2)	C14—C13—C20	117.07 (16)
C10—C1—H1	119.6	C15—C14—C19	117.8 (2)
C2—C1—H1	119.6	C15—C14—C13	122.11 (19)
C3—C2—C1	120.7 (3)	C19—C14—C13	120.05 (19)
C3—C2—H2	119.7	C16—C15—C14	120.5 (2)
C1—C2—H2	119.7	C16—C15—H15	119.7
C2—C3—C4	120.9 (2)	C14—C15—H15	119.7
C2—C3—H3	119.6	C17—C16—C15	120.4 (2)
C4—C3—H3	119.6	C17—C16—H16	119.8
C3—C4—C5	122.8 (2)	C15—C16—H16	119.8
C3—C4—C9	119.0 (2)	C18—C17—C16	119.8 (2)
C5—C4—C9	118.2 (3)	C18—C17—H17	120.1
C6—C5—C4	121.1 (3)	C16—C17—H17	120.1
C6—C5—H5	119.5	C17—C18—C19	120.4 (2)
C4—C5—H5	119.5	C17—C18—H18	119.8
C5—C6—C7	120.9 (3)	C19—C18—H18	119.8
C5—C6—H6	119.6	C18—C19—C14	121.0 (2)
C7—C6—H6	119.6	C18—C19—H19	119.5
C8—C7—C6	119.9 (3)	C14—C19—H19	119.5
C8—C7—H7	120.0	C25—C20—C21	118.2 (2)
C6—C7—H7	120.0	C25—C20—C13	121.51 (17)
C7—C8—C9	121.1 (2)	C21—C20—C13	120.2 (2)
C7—C8—H8	119.4	C22—C21—C20	120.8 (2)
C9—C8—H8	119.4	C22—C21—H21	119.6
C8—C9—C4	118.7 (2)	C20—C21—H21	119.6
C8—C9—C10	122.8 (2)	C23—C22—C21	120.2 (2)

C4—C9—C10	118.5 (2)	C23—C22—H22	119.9
C1—C10—N1	123.73 (19)	C21—C22—H22	119.9
C1—C10—C9	120.2 (2)	C22—C23—C24	119.9 (2)
N1—C10—C9	115.94 (19)	C22—C23—H23	120.1
N1—C11—C12	118.9 (2)	C24—C23—H23	120.1
N1—C11—H11	120.6	C23—C24—C25	120.2 (2)
C12—C11—H11	120.6	C23—C24—H24	119.9
C13—C12—C11	128.3 (2)	C25—C24—H24	119.9
C13—C12—H12	115.8	C24—C25—C20	120.7 (2)
C11—C12—H12	115.8	C24—C25—H25	119.7
C12—C13—C14	124.21 (19)	C20—C25—H25	119.7
C10—C1—C2—C3	0.5 (4)	C11—C12—C13—C20	−176.53 (18)
C1—C2—C3—C4	0.5 (4)	C12—C13—C14—C15	−49.7 (3)
C2—C3—C4—C5	179.2 (2)	C20—C13—C14—C15	126.6 (2)
C2—C3—C4—C9	−1.3 (3)	C12—C13—C14—C19	132.6 (2)
C3—C4—C5—C6	178.1 (2)	C20—C13—C14—C19	−51.1 (3)
C9—C4—C5—C6	−1.4 (4)	C19—C14—C15—C16	−0.4 (3)
C4—C5—C6—C7	0.0 (4)	C13—C14—C15—C16	−178.12 (19)
C5—C6—C7—C8	0.5 (4)	C14—C15—C16—C17	−0.2 (3)
C6—C7—C8—C9	0.4 (4)	C15—C16—C17—C18	0.4 (4)
C7—C8—C9—C4	−1.7 (3)	C16—C17—C18—C19	−0.2 (4)
C7—C8—C9—C10	179.8 (2)	C17—C18—C19—C14	−0.3 (4)
C3—C4—C9—C8	−177.3 (2)	C15—C14—C19—C18	0.6 (3)
C5—C4—C9—C8	2.2 (3)	C13—C14—C19—C18	178.42 (19)
C3—C4—C9—C10	1.2 (3)	C12—C13—C20—C25	−38.9 (3)
C5—C4—C9—C10	−179.31 (19)	C14—C13—C20—C25	144.7 (2)
C2—C1—C10—N1	−177.1 (2)	C12—C13—C20—C21	137.5 (2)
C2—C1—C10—C9	−0.6 (3)	C14—C13—C20—C21	−38.9 (3)
C11—N1—C10—C1	−40.3 (3)	C25—C20—C21—C22	2.3 (3)
C11—N1—C10—C9	143.1 (2)	C13—C20—C21—C22	−174.3 (2)
C8—C9—C10—C1	178.2 (2)	C20—C21—C22—C23	−1.4 (4)
C4—C9—C10—C1	−0.2 (3)	C21—C22—C23—C24	−0.5 (4)
C8—C9—C10—N1	−5.0 (3)	C22—C23—C24—C25	1.4 (4)
C4—C9—C10—N1	176.55 (18)	C23—C24—C25—C20	−0.5 (3)
C10—N1—C11—C12	−179.19 (18)	C21—C20—C25—C24	−1.3 (3)
N1—C11—C12—C13	171.0 (2)	C13—C20—C25—C24	175.22 (19)
C11—C12—C13—C14	−0.3 (3)		